

4-[[*(1E)*-(3,5-Dichloro-2-hydroxyphenyl)-methylene]amino]-1,5-dimethyl-2-phenyl-3*H*-pyrazol-3(2*H*)-one

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The crystal structure of the title compound, $C_{18}H_{15}Cl_2N_3O_2$, shows a strong intramolecular $O-H \cdots N$ hydrogen bond [$N \cdots O = 2.598(3) \text{ \AA}$, $O-H = 0.81(3) \text{ \AA}$, $H \cdots N = 1.86(3) \text{ \AA}$ and $O-H \cdots N = 152(3)^\circ$], which leads to the existence of a phenol-imine tautomer.

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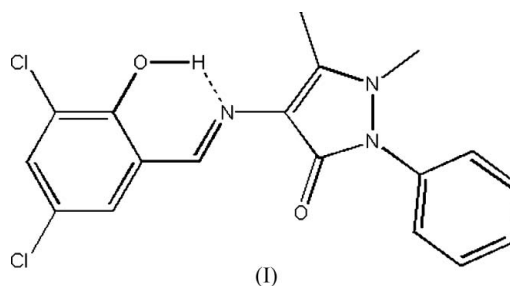
Key indicators

Single-crystal X-ray study
 $T = 293 \text{ K}$
Mean $\sigma(C-C) = 0.004 \text{ \AA}$
 R factor = 0.047
 wR factor = 0.135
Data-to-parameter ratio = 12.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Comment

Great interest has been devoted to the preparation and study of the Schiff bases derived from salicylaldehyde due to their tautomeric structure (Salman *et al.*, 1991), fluorescent (Morishige *et al.*, 1980), and thermo- and photochromic properties (Barbara *et al.*, 1980; Cohen *et al.*, 1964). In a search for new analytical reagents, we have synthesized some compounds of substituted salicylaldehyde with 4-aminoantipyrene (Huang *et al.*, 2005). We report here the synthesis and crystal structure of the title compound, (I).



All the bond distances and angles are normal and agree with the corresponding values found in a similar compound, *viz.* 4-[(2-hydroxy-3-methoxybenzylidene)amino]-1,5-dimethyl-2-phenyl-1*H*-pyrazol-3(2*H*)-one (Diao *et al.*, 2005). There is an intramolecular $O-H \cdots N$ hydrogen bond (Table 2); the compound is in the phenol-imine form, as in 4-[[*(1E)*-(2-hydroxyphenyl)methylidene]amino]-1,5-dimethyl-2-phenyl-2,3-dihydro-1*H*-pyrazol-3-one [$N1 \cdots O1 = 2.607(3) \text{ \AA}$, $O1-H1 = 0.97(3) \text{ \AA}$, $H1 \cdots N1 = 1.71(3) \text{ \AA}$ and $O1-H1 \cdots N1 = 153(2)^\circ$; Hökelek *et al.*, 2001].

Experimental

3,5-Dichlorosalicylaldehyde was prepared according to the method of Sukuzi & Takashi (1983). Ethanol solutions of 3,5-dichlorosalicylaldehyde (10 mmol, 1.70 g) and 4-aminoantipyrene (10 mmol, 2.03 g) were mixed and refluxed on a water bath for 2 h. After cooling, the separated precipitate was filtered off, washed and recrystallized from methanol (yield: 83%; m.p. 498.6–499.1 K). IR (KBr, cm^{-1}): ν_{max} 3430.5, 1664.5, 1592.1, 1452.3, 1356.8, 1290.3, 1136.0, 766.7. $^1\text{H NMR}$ (200 MHz, CDCl_3): δ 14.25 (1H), 9.72 (1H), 7.18–7.59 (7H), 3.21 (3H), 2.42 (3H).

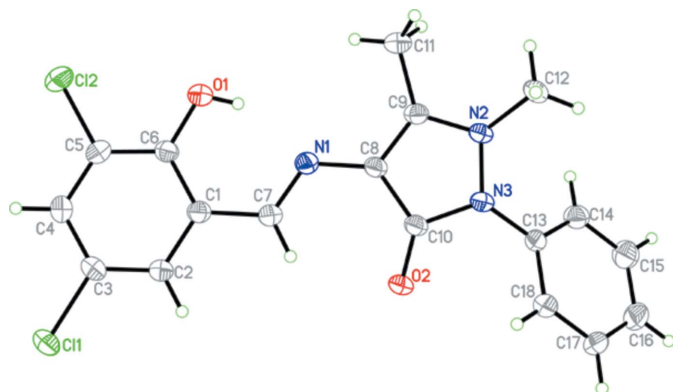


Figure 1
The structure of compound (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.

Crystal data

$C_{18}H_{15}Cl_2N_3O_2$
 $M_r = 376.23$
 Monoclinic, $P2_1/n$
 $a = 7.0146$ (6) Å
 $b = 8.0466$ (7) Å
 $c = 30.510$ (3) Å
 $\beta = 90.921$ (2)°
 $V = 1721.9$ (3) Å³
 $Z = 4$

$D_x = 1.451$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 3109 reflections
 $\theta = 1.3$ – 25.3 °
 $\mu = 0.39$ mm⁻¹
 $T = 293$ (2) K
 Block, orange
 $0.22 \times 0.17 \times 0.16$ mm

Data collection

Siemens SMART CCD area-detector diffractometer
 ω and φ scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{min} = 0.923$, $T_{max} = 0.939$
 8818 measured reflections

3109 independent reflections
 2510 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.031$
 $\theta_{max} = 25.3$ °
 $h = -6 \rightarrow 8$
 $k = -9 \rightarrow 9$
 $l = -36 \rightarrow 35$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.136$
 $S = 1.14$
 3109 reflections
 255 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.067P)^2 + 0.3379P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} = 0.001$
 $\Delta\rho_{max} = 0.33$ e Å⁻³
 $\Delta\rho_{min} = -0.26$ e Å⁻³
 Extinction correction: SHELXL97
 Extinction coefficient: 0.0028 (13)

Table 1

Selected geometric parameters (Å, °).

| | | | |
|----------|-----------|----------|-----------|
| C11—C3 | 1.744 (3) | N1—C8 | 1.395 (3) |
| C12—C5 | 1.736 (3) | N2—N3 | 1.412 (3) |
| O1—C6 | 1.340 (3) | N3—C10 | 1.403 (3) |
| O2—C10 | 1.233 (3) | | |
| C7—N1—C8 | 120.7 (2) | O1—C6—C5 | 120.4 (2) |
| C2—C3—C4 | 121.3 (2) | O1—C6—C1 | 121.3 (2) |

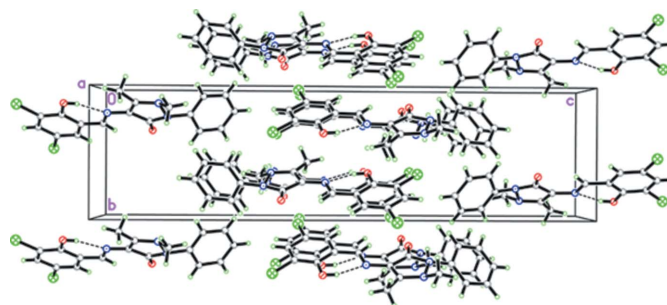


Figure 2
The packing diagram of (I), viewed down the a axis. Dashed lines indicate hydrogen bonds.

Table 2

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--------------------|----------|-------------|-------------|---------------|
| O1—H1A \cdots N1 | 0.81 (3) | 1.86 (3) | 2.598 (3) | 152 (3) |

The hydroxy H atom (H1A) and the methyl H atom were positioned from a difference map, refined several cycles then fixed at a distance of 0.80 Å; the methyl H atoms on C11 and C12 were located in a Fourier synthesis and refined freely. The remaining H atoms were positioned geometrically and treated as riding, at distances of 0.93 (CH) and 0.96 Å (CH₃) and with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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References

- Barbara, P. F., Rentzepis, P. M. & Brus, L. E. (1980). *J. Am. Chem. Soc.* **102**, 2786–2791.
 Bruker (2002). SMART (Version 5.62), SAINT (Version 6.02) and SADABS (Version 2.03). Bruker AXS Inc., Madison, Wisconsin, USA.
 Cohen, M. D., Schmidt, G. M. J. & Flavin, S. (1964). *J. Chem. Soc.* pp. 2041–2051.
 Diao, C.-H., Fan, Z. & Yu, M. (2005). *Acta Cryst.* **E61**, o3271–o3272.
 Hökelek, T., Işıklan, M. & Kılıç, Z. (2001). *Acta Cryst.* **C57**, 117–119.
 Huang, L. & Chen, D.-B. (2005). *Acta Cryst.* **E61**, o4169–o4170.
 Morishige, K. (1980). *Anal. Chim. Acta*, **121**, 301–308.
 Salman, S. R., Farrant, R. D. & Lindon, J. C. (1991). *Spectrosc. Lett.* **24**, 1071–1078.
 Sheldrick, G. M. (1996). SHELXL93. University of Göttingen, Germany.
 Sheldrick, G. M. (1997a). SHELXL97 and SHELXS97. University of Göttingen, Germany.
 Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
 Sukuzi, Y. & Takashi, H. (1983). *Chem. Pharm. Bull.* **31**, 1751–1753.